# Thin Film Behavior after Ink Transfer in Printing Processes



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# <u>Overview</u> Thin Film Behavior after Ink Transfer in Printing Processes



### Motivation

- Graphic vs. functional printing
- Printing processes for organic electronics, challenges
- Film formation process in R2R: Process chain

### Theory

- Navier-Stokes in the lubrication limit: The Landau-Levich equation
- Effects of surface tension and concentration gradients
- Stability analysis: Phase diagrams

# Stability analysis

- Constant surface tension
- Why is a puddle stationary flat?
- Additional forces



# <u>Motivation</u>

# Graphic vs. functional printing



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#### **GRAPHIC printing:**

- blue ink gravure printed on PET,
- 1.2mm x 0.9mm, height ~ 4µm

# ⇒ HOMOGENOUS, DEFINED DOT SCREENS



### **FUNCTIONAL printing:**

- SY organic polymer for OLEDs, gravure printed on PET,
- 240µm x 180µm, height ~ 30nm

# $\Rightarrow$ HOMOGENOUS, DEFINED CLOSED LAYERS



# <u>Motivation</u>

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Printing process for organic electronics: Challenges

# Thin and homogenous layers i.e. OLEDs:

- dewetting: rupture, holes
- crystallization

### Multilayer devices:

- compatibility of material sets
- stability of under-laying film
- diffusion of liquid or solutes into under-laying film
- register accuracy

### Multi-component fluids:

- different solutes: polymers and/or small molecules
- different liquids: water-based and/or solvent-based solution



240µm x 180µm, height ~100nm

**P3HT on PET** 



# <u>Motivation</u>



# Film formation in R2R: Process chain





# <u>Theory</u> Navier-Stokes in the lubrication limit: The Landau Levich equation









Small perturbed liquid film, leveling time:

$$\tau \propto \frac{\sigma h_0^{-3}}{\eta \lambda^4}$$

[2] L. Landau, B. Levich, Acta Physicochim. URSS. **1942**, 42-54
[3] A. Oron, S.G. Bankoff, Rev. Mod. Phys. **1997**, 69, 931-980
[4] P. de Gennes, Rev. Mod. Phys. **1985**, 57, 827-863.



### <u>Theory</u>



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# Effects of surface tension and concentration gradients



Thin liquid film of a binary system:

- C : concentration of solute in solution
- $\sigma$  : surface tension
- $\lambda_{v}$  diffusion length

Useful relation [1]:

$$\frac{\partial C}{\partial z}\Big|_{z=0} = \frac{\lambda_D}{2k_B T} \frac{\partial \sigma_{sf}}{\partial C}$$

In the following: Evaporation and temperature gradients effects are neglected.

[1] J.W. Cahn, J. Chem. Phys. 1977, 66, 3667



### <u>Theory</u>

# Stability analysis: Phase diagrams



When do we have solutions of L.L. eq. (1) for <u>stable</u>, homogenous flat, large-scale films concerning  $\underline{C}, \underline{\sigma}$ <u>gradients</u> and <u>Van der Waals</u> forces?

STABLE: 
$$\frac{\partial h(x,t)}{\partial t} = 0 \implies \text{eq.}(1) \rightarrow h'(h)$$
  
GRADIENTS of  $C, \sigma : \rightarrow \beta$   
VAN DER WAALS :  $\rightarrow A \implies h'(h)|_{\beta, A, c_0}$   
INTEGRATION CONSTANT:  $c_0$ 



# **Constant surface tension**



No additional forces:  $\partial \sigma / \partial x = 0$  and A = 0







# Why is a puddle stationary flat?



No additional forces:  $\partial \sigma / \partial x = 0$  and A = 0







# Why is a puddle stationary flat?



No additional forces:  $\partial \sigma / \partial x = 0$  and A = 0



![](_page_10_Picture_6.jpeg)

# **Additional forces**

![](_page_11_Picture_2.jpeg)

Gradients in C ,  $\sigma$ :  $\beta = 2$  and Van der Waals forces: A = 2

![](_page_11_Figure_4.jpeg)

![](_page_11_Picture_5.jpeg)

![](_page_11_Picture_6.jpeg)

![](_page_12_Picture_0.jpeg)

![](_page_12_Picture_1.jpeg)

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![](_page_12_Picture_6.jpeg)

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